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*CdTe*,

**STRUCTURAL AND PHASE CHANGES IN A LEAD TELLURIDE MATRIX WITH ADDITIVES OF CADMIUM SULPHIDE**

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***Abstract.** Structural and phase changes in lead telluride with cadmium sulfide additives have been studied. It has been shown that the crystal structures and lattice parameters of the molecules forming new phases change. The higher the concentration of these impurity atoms, the larger the lattice parameter *a* for the CdTe phase with the sphalerite structure. As the concentration of cadmium sulfide increases, new three-component phases without lead content are formed, which can be represented as a solid solution with variable composition.*

**Keywords:** lead telluride, thermoelectric material, cadmium sulfide, crystal cell, X-ray phase analysis, lattice parameter

[1-4].

(PbTe)

$A^{IV}B^{VI}$ ,

[5]. Pb-Te (924 ° )  
 Te (50,002 %).

(NaCl).  
(CdS)

$A^{II}B^{VI}$ ,

$A^{II}B^{VI}$   
[6-8].

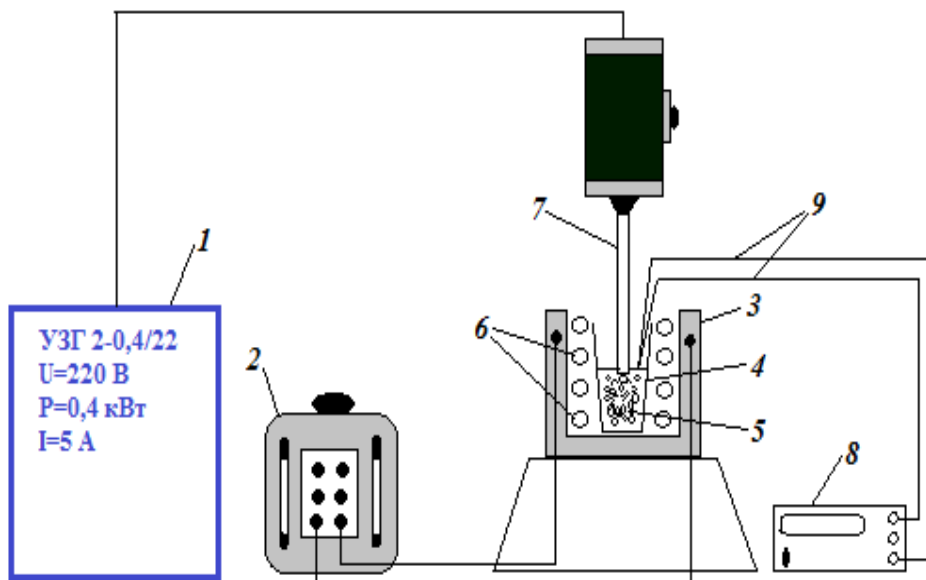
PbTe

CdS

PbTe

CdS

1.  
(1, 3, 5, 8 10 %).  
(99,999 %), (99,999 %)



1 - ; 2 - ; 3 - ; 4 - ; 5 - ;  
6 - ; 7 - ; 8 - ; 9 -

[9, 10].

1183 .  
400 .

15 22 1203 . 130 /  
823 , 750 . 8 .  
10

-6

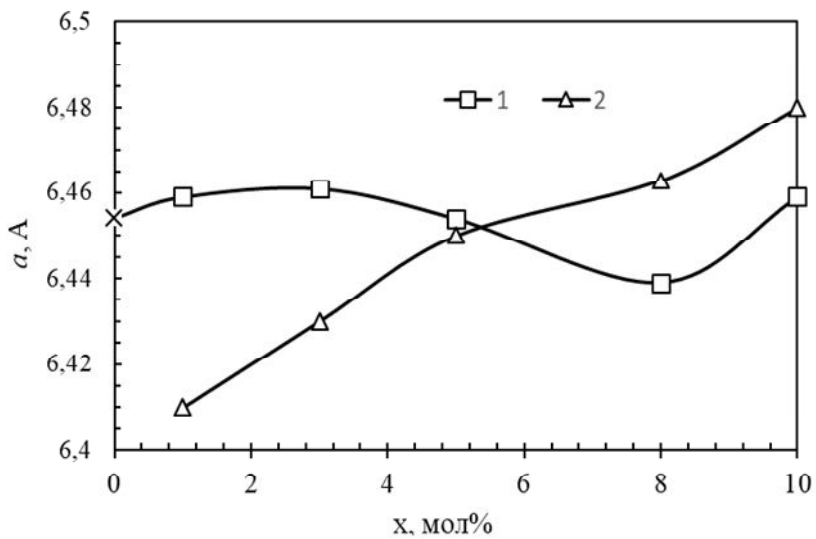
« CdS PbTe » [11].  
1.

1 -

CdS PbTe, %						
0	PbTe	.	Fm-3m	$a=6,45900$	269,46	8,253
1	PbTe	.	Fm-3m	$a=6,45900$	269,46	8,253
	$(\text{Cd}_{0,16}\text{Pb}_{0,84})\text{Te}$	.	Fm-3m	$a=6,45400$	268,84	8,272
	CdTe	.	F-43m	$a=6,41000$	278,45	5,725
	$\text{Pb}(\text{S}_{0,6}\text{Te}_{0,4})$	.	Fm-3m	$a=6,46000$	269,59	8,249
3	PbTe	.	Fm-3m	$a=6,46100$	269,71	8,245
	$\text{Cd}(\text{S}_{0,6}\text{Te}_{0,4})$	.	P63mc	$a=4,40430$ $c=7,20530$	121,04	5,784
	CdTe	.	F-43m	$a=6,43000$	278,45	5,725
	$(\text{Cd}_{0,06}\text{Pb}_{0,94})\text{S}$	.	Fm-3m	$a=6,10500$	117,87	5,803
5	PbTe	.	Fm-3m	$a=6,45410$	268,85	8,245
	$\text{CdS}_{0,6}\text{Te}_{0,4}$	.	F-43m	$a=6,43140$	266,02	5,871
	CdTe	.	Fm-3m	$a=6,45000$	263,37	6,053
8	CdTe	.	Fm-3m	$a=6,46300$	263,37	6,053
	PbTe	.	Fm-3m	$a=6,43900$	266,97	8,150
	$\text{CdS}_{0,9}\text{Te}_{0,1}$	.	F-43m	$a=6,46140$	266,02	5,871
10	PbTe	.	Fm-3m	$a=6,45900$	269,46	8,190
	CdTe	.	F-43m	$a=6,48000$	272,10	5,859
	$\text{Cd}(\text{S}_{0,6}\text{Te}_{0,4})$	.	P63mc	$a=4,42430$ $=7,20530$	121,04	5,784
	$\text{Cd}(\text{S}_{0,7}\text{Te}_{0,3})$	.	P63mc	$a=4,36000$ $=7,16000$	117,87	5,803

CdS
1 %
CdTe
PbTe

[12].  
 ( 2).



2 - CdS PbTe (1) CdTe (2)  
PbTe-CdS

